

# A Comparison of (*R,R*)-Me-DUPHOS and (*R,R*)-DUPHOS-*i*Pr Ligands in the Pd<sup>0</sup>-Catalysed Asymmetric Allylic Alkylation Reaction: Stereochemical and Kinetic Considerations

Vanda Raquel Marinho,<sup>[a]</sup> J. P. Prates Ramalho,<sup>[a]</sup> Ana I. Rodrigues,<sup>[b]</sup> and Anthony J. Burke<sup>\*[a]</sup>

**Keywords:** Asymmetric catalysis / Alkylation / Palladium / P ligands / Fukui function / Kinetics

It has come to our attention that the absolute configuration shown for (*R,R*)-DUPHOS-*i*Pr (**1b**) shown in Figure 1 of the original article,<sup>[1]</sup> was in fact the (*S,S*) configuration and not the purported (*R,R*) absolute configuration. This is a consequence of the subtlety of the Cahn–Ingold–Prelog rules, a point that had previously been made by Burk et al.,<sup>[2]</sup> but had unfortunately been overlooked by us at the time of writing and of submitting this paper. Thus, Figure 1 should be as shown below.

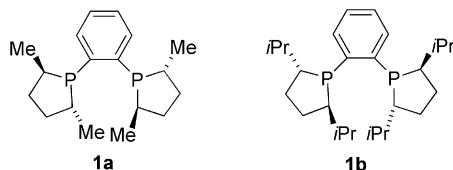


Figure 1. (*R,R*)-Me-DUPHOS (**1a**) and (*R,R*)-DUPHOS-*i*Pr (**1b**).

This therefore means that, in fact, complexes **2b** and **2c**, have the opposite absolute configuration to that shown as Figure 2 in our original paper (see the corrected version for Figure 2 below), and since the calculations on the Fukui function were inadvertently performed on the complex with the opposite absolute configuration to that used in the experiments (see Figure 4 in our original paper for the picture of the calculated structure), the preferred site of attack (C1) on the  $\pi$ -allyl complex **2b** (indicated as C1 in the corrected Figure 2) will give the malonate product with the (*R*) absolute configuration, as observed. Gratifyingly, this now explains on a very simple basis, the switch in the absolute configuration of the malonate product from (*S*) to (*R*) on going from (*R,R*)-DUPHOS-Me to (*R,R*)-DUPHOS-*i*Pr, and shows that, in both cases, it is the electronic effect and not any stereochemical effect, that is the controlling influence.

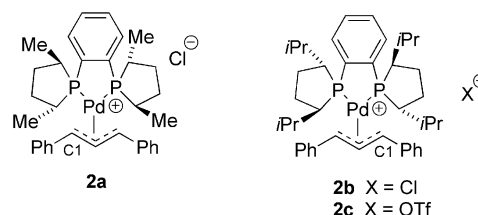


Figure 2. Pd-allyl complexes **2**.

In our original paper, both the models shown in Figures 5 and 6 are incorrect as they depict the calculated structures of the complexes formed from the (*S,S*) enantiomer of DUPHOS-*i*Pr. These calculations now become redundant, as the switch in absolute configuration has now been explained (see above). Scheme 2 is also incorrect, as it depicts the wrong enantiomer of **2b**, but it suffices to give the general idea of the possible interconversions involved.

Figure 8 in our original paper should also be corrected as shown in the following Figure 8.

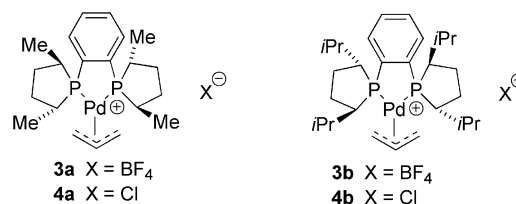


Figure 8. Isolated Pd-allyl complexes **3a** and **3b** formed using AgBF<sub>4</sub> and in situ formed complexes **4a** and **4b**.

The authors wish to apologise for any inconvenience this may have caused readers of our paper.

[a] Departamento de Química and Centro de Química de Évora, Universidade de Évora, Rua Romão Romalho 59, 7000 Évora, Portugal  
Fax: +351-266745303  
E-mail: ajb@dquim.uevora.pt

[b] Departamento de Tecnologia de Indústrias Químicas, Instituto Nacional de Engenharia, Tecnologia e Inovação, Estrada do Paço do Lumiar, Edifício F, 1649-038 Lisboa, Portugal

[1] V. R. Marinho, J. P. P. Ramalho, A. I. Rodrigues, A. J. Burke, *Eur. J. Org. Chem.* **2009**, 6311–6317.

[2] M. J. Burk, J. E. Feaster, W. A. Nugent, R. L. Harlow, *J. Am. Chem. Soc.* **1993**, *115*, 10125–10138.

Received: January 14, 2010  
Published Online: January 29, 2010